

## LAMPIRAN

Lampiran 1.

No.	Sampel	Absorbansi sampel		Absorbansi rata-rata
1.	A	0,425	0,517	0,471
2.	B	0,727	0,747	0,737
3.	C	0,350	0,337	0,343
4.	D	0,363	0,296	0,329
5.	E	0,484	0,473	0,478
6.	F	0,490	0,536	0,513
7.	G	0,800	0,713	0,756

### Konsentrasi asam benzoat dalam sampel

- Sampel A

Diketahui: persamaan regresi  $y = 0,0203x + 0,1894$   
 $Y = 0,471$

$$Y = bx + a$$

$$Y = 0,0203x + 0,1894$$

$$0,471 = 0,0203x + 0,1894$$

$$x = \frac{0,471 - 0,1894}{0,0203}$$

$$= 13,871 \text{ ppm}$$

- Sampel B

Diketahui: persamaan regresi  $y = 0,0203x + 0,1894$   
 $Y = 0,737$

$$Y = bx + a$$

$$Y = 0,0203x + 0,1894$$

$$0,737 = 0,0203x + 0,1894$$

$$x = \frac{0,737 - 0,1894}{0,0203}$$

$$= 26,975 \text{ ppm}$$

- Sampel C

Diketahui: persamaan regresi  $y = 0,0203x + 0,1894$   
 $Y = 0,343$

$$Y = bx + a$$

$$Y = 0,0203x + 0,1894$$

$$0,343 = 0,0203x + 0,1894$$

$$x = \frac{0,343 - 0,1894}{0,0203}$$

$$= 7,566 \text{ ppm}$$

- Sampel D

Diketahui: persamaan regresi  $y = 0,0203x + 0,1894$   
 $Y = 0,329$

$$\begin{aligned} Y &= bx + a \\ Y &= 0,0203x + 0,1894 \\ 0,329 &= 0,0203x + 0,1894 \\ x &= \frac{0,329 - 0,1894}{0,0203} \\ &= 6,876 \text{ ppm} \end{aligned}$$

- Sampel E

Diketahui: persamaan regresi  $y = 0,0203x + 0,1894$   
 $Y = 0,478$

$$\begin{aligned} Y &= bx + a \\ Y &= 0,0203x + 0,1894 \\ 0,478 &= 0,0203x + 0,1894 \\ x &= \frac{0,478 - 0,1894}{0,0203} \\ &= 14,216 \text{ ppm} \end{aligned}$$

- Sampel F

Diketahui: persamaan regresi  $y = 0,0203x + 0,1894$   
 $Y = 0,513$

$$\begin{aligned} Y &= bx + a \\ Y &= 0,0203x + 0,1894 \\ 0,513 &= 0,0203x + 0,1894 \\ x &= \frac{0,513 - 0,1894}{0,0203} \\ &= 15,940 \text{ ppm} \end{aligned}$$

- Sampel G

Diketahui: persamaan regresi  $y = 0,0203x + 0,1894$   
 $Y = 0,756$

$$\begin{aligned} Y &= bx + a \\ Y &= 0,0203x + 0,1894 \\ 0,756 &= 0,0203x + 0,1894 \\ x &= \frac{0,756 - 0,1894}{0,0203} \\ &= 27,911 \text{ ppm} \end{aligned}$$

**Kadar asam benzoat dalam sampel**

- Sampel A

Diketahui:  $C = 13,871 \text{ mg/L}$   
 $V = 0,1 \text{ L}$   
 $Fp = 100/25 = 4$

$$\begin{aligned} \text{Kadar} &= \frac{C \times V \times Fp}{W} \\ &= \frac{13,871 \text{ mg/l} \times 0,1 \text{ L} \times 4}{0,005 \text{ kg}} \\ &= 1.109 \text{ mg/kg} \\ &= 1,109 \text{ g/kg} \end{aligned}$$

- Sampel B

Diketahui:  $C = 26,975 \text{ mg/L}$   
 $V = 0,1 \text{ L}$   
 $Fp = 100/25 = 4$

$$\begin{aligned} \text{Kadar} &= \frac{C \times V \times Fp}{W} \\ &= \frac{26,975 \text{ mg/l} \times 0,1 \text{ L} \times 4}{0,005 \text{ kg}} \\ &= 2.158 \text{ mg/kg} \\ &= 2,158 \text{ g/kg} \end{aligned}$$

- Sampel C

Diketahui:  $C = 7,566 \text{ mg/L}$   
 $V = 0,1 \text{ L}$   
 $Fp = 100/25 = 4$

$$\begin{aligned} \text{Kadar} &= \frac{C \times V \times Fp}{W} \\ &= \frac{7,566 \text{ mg/l} \times 0,1 \text{ L} \times 4}{0,005 \text{ kg}} \\ &= 605 \text{ mg/kg} \\ &= 0,605 \text{ g/kg} \end{aligned}$$

- Sampel D

Diketahui:  $C = 6,876 \text{ mg/L}$   
 $V = 0,1 \text{ L}$   
 $Fp = 100/25 = 4$

$$\begin{aligned} \text{Kadar} &= \frac{C \times V \times Fp}{W} \\ &= \frac{6,876 \text{ mg/l} \times 0,1 \text{ L} \times 4}{0,005 \text{ kg}} \\ &= 550 \text{ mg/kg} \\ &= 0,550 \text{ g/kg} \end{aligned}$$

- Sampel E

Diketahui:  $C = 14,216 \text{ mg/L}$   
 $V = 0,1 \text{ L}$

$$F_p = 100/25 = 4$$

$$\begin{aligned} \text{Kadar} &= \frac{C \times V \times F_p}{W} \\ &= \frac{14,216 \text{ mg/l} \times 0,1 \text{ L} \times 4}{0,005 \text{ kg}} \\ &= 1.137 \text{ mg/kg} \\ &= 1,137 \text{ g/kg} \end{aligned}$$

- Sampel F  
Diketahui: C = 15,940 mg/L  
V = 0,1 L  
F<sub>p</sub> = 100/25 = 4

$$\begin{aligned} \text{Kadar} &= \frac{C \times V \times F_p}{W} \\ &= \frac{15,940 \text{ mg/l} \times 0,1 \text{ L} \times 4}{0,005 \text{ kg}} \\ &= 1.275 \text{ mg/kg} \\ &= 1,275 \text{ g/kg} \end{aligned}$$

- Sampel G  
Diketahui: C = 27,911 mg/L  
V = 0,1 L  
F<sub>p</sub> = 100/25 = 4

$$\begin{aligned} \text{Kadar} &= \frac{C \times V \times F_p}{W} \\ &= \frac{27,911 \text{ mg/l} \times 0,1 \text{ L} \times 4}{0,005 \text{ kg}} \\ &= 2.232 \text{ mg/kg} \\ &= 2,232 \text{ g/kg} \end{aligned}$$

### **Kadar natrium benzoat dalam sampel**

- Sampel A  
Diketahui: Kadar Asam Benzoat dalam sampel = 1.109 mg/kg  
BM Natrium Benzoat = 144,11  
BM Asam Benzoat = 122,12

$$\begin{aligned} \text{Kadar} &= \text{kadar As. Benzoat} \times \frac{BM \text{ Na.Benzoat}}{BM \text{ As.Benzoat}} \\ &= 1.109 \text{ mg/kg} \times \frac{144,11}{122,12} \\ &= 1.308,62 \text{ mg/kg} \end{aligned}$$

- Sampel B  
 Diketahui: Kadar Asam Benzoat dalam sampel = 2.158 mg/kg  
                   BM Natrium Benzoat = 144,11  
                   BM Asam Benzoat = 122,12

$$\begin{aligned} \text{Kadar} &= \text{kadar As. Benzoat} \times \frac{\text{BM Na.Benzoat}}{\text{BM As.Benzoat}} \\ &= 2.158 \text{ mg/kg} \times \frac{144,11}{122,12} \\ &= 2.546,44 \text{ mg/kg} \end{aligned}$$

- Sampel C  
 Diketahui: Kadar Asam Benzoat dalam sampel = 605 mg/kg  
                   BM Natrium Benzoat = 144,11  
                   BM Asam Benzoat = 122,12

$$\begin{aligned} \text{Kadar} &= \text{kadar As. Benzoat} \times \frac{\text{BM Na.Benzoat}}{\text{BM As.Benzoat}} \\ &= 605 \text{ mg/kg} \times \frac{144,11}{122,12} \\ &= 713,9 \text{ mg/kg} \end{aligned}$$

- Sampel D  
 Diketahui: Kadar Asam Benzoat dalam sampel = 550 mg/kg  
                   BM Natrium Benzoat = 144,11  
                   BM Asam Benzoat = 122,12

$$\begin{aligned} \text{Kadar} &= \text{kadar As. Benzoat} \times \frac{\text{BM Na.Benzoat}}{\text{BM As.Benzoat}} \\ &= 550 \text{ mg/kg} \times \frac{144,11}{122,12} \\ &= 649 \text{ mg/kg} \end{aligned}$$

- Sampel E  
 Diketahui: Kadar Asam Benzoat dalam sampel = 1.137 mg/kg  
                   BM Natrium Benzoat = 144,11  
                   BM Asam Benzoat = 122,12

$$\begin{aligned} \text{Kadar} &= \text{kadar As. Benzoat} \times \frac{\text{BM Na.Benzoat}}{\text{BM As.Benzoat}} \\ &= 1.137 \text{ mg/kg} \times \frac{144,11}{122,12} \\ &= 1.341,66 \text{ mg/kg} \end{aligned}$$

- Sampel F
 

Diketahui:	Kadar Asam Benzoat dalam sampel	= 1.1275 mg/kg
	BM Natrium Benzoat	= 144,11
	BM Asam Benzoat	= 122,12

$$\begin{aligned}
 \text{Kadar} &= \text{kadar As. Benzoat} \times \frac{BM \text{ Na.Benzoat}}{BM \text{ As.Benzoat}} \\
 &= 1.275 \text{ mg/kg} \times \frac{144,11}{122,12} \\
 &= 1.504,5 \text{ mg/kg}
 \end{aligned}$$

- Sampel G
 

Diketahui:	Kadar Asam Benzoat dalam sampel	= 2.232 mg/kg
	BM Natrium Benzoat	= 144,11
	BM Asam Benzoat	= 122,12

$$\begin{aligned}
 \text{Kadar} &= \text{kadar As. Benzoat} \times \frac{BM \text{ Na.Benzoat}}{BM \text{ As.Benzoat}} \\
 &= 2.232 \text{ mg/kg} \times \frac{144,11}{122,12} \\
 &= 2.633,76 \text{ mg/kg}
 \end{aligned}$$